

## WHAT CAN YOU CONCLUDE ABOUT IDENTIFICATION OF COMPOUND PHYSICAL PROPERTIES%0A

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What can you conclude about the strength of ionic bonds based on the physical properties of ionic compounds? Based off the physical properties, of how the ions are packed together in a repeating pattern in which balance the forces of attraction and repulsion, I can conclude that there is a pretty decent strength between the bonds.

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### **Chemical Bonds and Compounds Flashcards Quizlet**

Created by ionic bonds. Ionic compounds are very stable and their crystals very strong. The shape of the crystals of an ionic compound depends on the ratio of positive and negative ions and the sizes of the ions.

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### **Physical Chemical Properties of Compounds Types**

There are many types of physical properties that can be used to tell compounds apart. Let's look at some of the more common physical properties of compounds. Color and odor are physical properties

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### **Identifying an Unknown Compound by Solubility Functional**

Many compounds have similar physical properties and give similar results in qualitative tests. However, an unknown can undergo reaction to form another compound called a derivative. The melting point of the purified derivative allows identification of the unknown. In this lab we will focus on using Solubility Tests, Chemical Tests and Spectra Analysis to identify two unknown compounds

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### **Identify Physical and Chemical Changes Physical and**

Identify each term as either Chemical Change, Physical Change, Chemical Property, or Physical Property

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### **Compounds and Mixtures Quiz Date Name BrainPOP**

Compounds and Mixtures Quiz 1. How is a compound different from a mixture? a. Mixtures are created through physical changes; compounds are created through chemical reactions.

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### **Questions A Why are biochemical tests used to identify**

Questions A. Why are biochemical tests used to identify microbes? Most other organisms are

analyzed and characterized by their physical appearances through the naked eye, however you are unable to do this with microbes so biochemical identification is used instead.

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### **Matter Properties and Changes**

properties can be observed without changing a substance's chemical composition. Color, hardness, and (5) are examples. Other properties cannot be observed without changing the composition of a substance. These are called (6) properties. An example is the tendency of iron to form rust when exposed to air. Label each property as either physical or chemical. 7. Chemical formula H<sub>2</sub>O 8. Forms

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### **EXPERIMENT 3 Identification of a Substance by Physical**

can be classified as either physical properties or chemical properties. Physical properties are those that can be Physical properties are those that can be determined or measured without changing the composition or identity of the substance.

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### **Experiment 1 Identification of a Compound Physical Properties**

Experiment 1 Identification of a Compound: Physical Properties I NTRODUCTION Gold is yellow, roses are often red, salt is white and crystalline, lead is "heavy," and water is

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### **Identification of Substances by Physical Properties**

Physical properties are those properties that can be observed without altering the composition of the substance. Some physical properties that can be observed quantitatively are melting point, boiling point, solubility, viscosity, density, and refractive index.

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### **Identification of a Substance by Physical Properties**

that can be determined or measured without changing the composition or identity of the substance. These These properties include color, odor, taste, density, melting point, boiling point, conductivity, and hardness.

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### **Experiment3 Identification of a Compound Chemical**

Experiment 2 Identification of a Compound: Chemical Properties Introduction Chemists and scientists in general, develop and design experiments in an attempt

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